

STRUCTURES OF MARCFORTINE B AND C (X-RAY ANALYSIS), ALKALOIDS  
FROM PENICILLIUM ROQUEFORTI<sup>§</sup>

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Summary : Marcfortine B and C are minor alkaloids isolated from the mycelium of Penicillium roqueforti. The structure of marcfortine B was established by spectral means and that of marcfortine C by X-ray diffraction analysis.

A previous study <sup>1</sup> of the Penicillium roqueforti strain B26 resulted in the isolation, along with roquefortine <sup>2</sup>, of three new alkaloids which have been designated marcfortine A, B and C. The major component, marcfortine A,  $C_{28}H_{35}N_3O_4$ , was shown to have structure 1 and to be the first fungal alkaloid to possess a seven membered ring formed by linkage of a isoprene unit to two phenolic hydroxy-groups on the tryptophan unit. We herein report the structural elucidation of the two other components, marcfortine B 2 and marcfortine C 3.

Marcfortine B 2 crystallized from ethyl acetate, m.p. 178-180°,  $[\alpha]_D^{22} - 67.7^\circ$  (c 1.77,  $CHCl_3$ ) ; u.v.  $\lambda_{max}$  229 nm ( $\epsilon$  20755). The high resolution mass spectrum established the molecular formula of 2 as  $C_{27}H_{33}N_3O_4$  ( $M^+$  at m/z 463.2479) which differed from that of marcfortine A by one methylene unit. This difference is accounted for by the absence of a resonance due to a N-CH<sub>3</sub> grouping in the <sup>1</sup>H-n.m.r. spectrum of 2 and suggests that 2 was 29-nor-marcfortine A. The 250 MHz <sup>1</sup>H n.m.r. of marcfortine B (Table I) displayed in the downfield region the same resonances as that of 1, namely the AB quartet due to the two contiguous aromatic protons (H-4 and H-5) and the widely separated olefinic AB quartet due to H-24 and H-25. The chemical shifts for the four methyl groups were also identical whereas the pattern between  $\delta$  1.5 and 3.7 ppm was slightly different. Double resonance experiments identified other signals : the triplet due to H-20 and the three two-proton AB patterns due to the methylenes  $\alpha$  to nitrogen (H-12 and H-17) and to the C-10 methylene. The <sup>13</sup>C n.m.r. spectrum (Table II) of marcfortine B 2 was also

<sup>§</sup>This paper is dedicated to the memory of the late Professor F. Šorm.

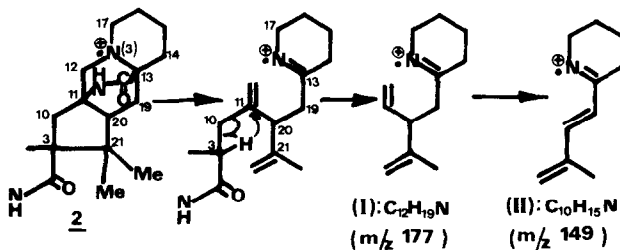
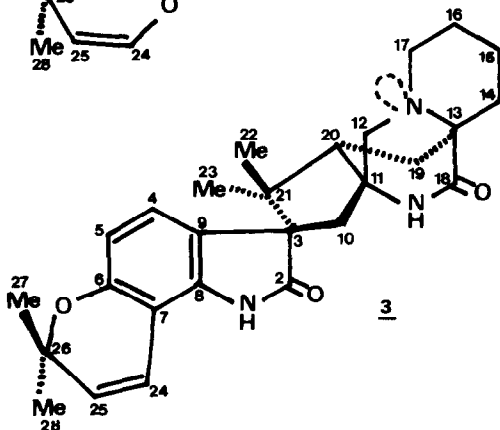
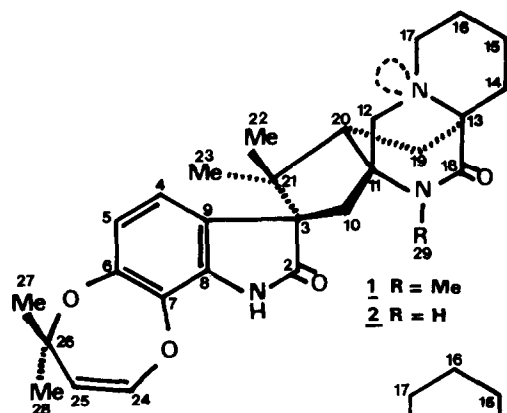
Table I : 250 MHz  $^1\text{H}$  n.m.r. spectra of 2 and 3 [ $\delta$  in ppm, J as (Hz)].

	<u>2</u>	<u>3</u>
H-4	6.66 d (8.1)	6.86 d (8.1)
H-5	6.79 d	6.42 d
H-24	6.39 d (7.5)	6.44 d (10)
H-25	4.91 d	5.82 d
H-12	3.71 (11.2)	3.68 d (11.2)
	2.48 br.d	2.45 br.d
H-20	3.08 t (10)	3.08 t (10)
H-17	2.70 dd (11.2)	2.67 dd (11.2)
	2.48 br.d	2.45 br.d
26-Me	1.45 1.43	1.45 1.43
21-Me	1.08 0.81	1.08 0.81

Table II :  $^{13}\text{C}$  n.m.r. spectra <sup>3</sup> of 1, 2 and 3; a-d signals may be reversed.

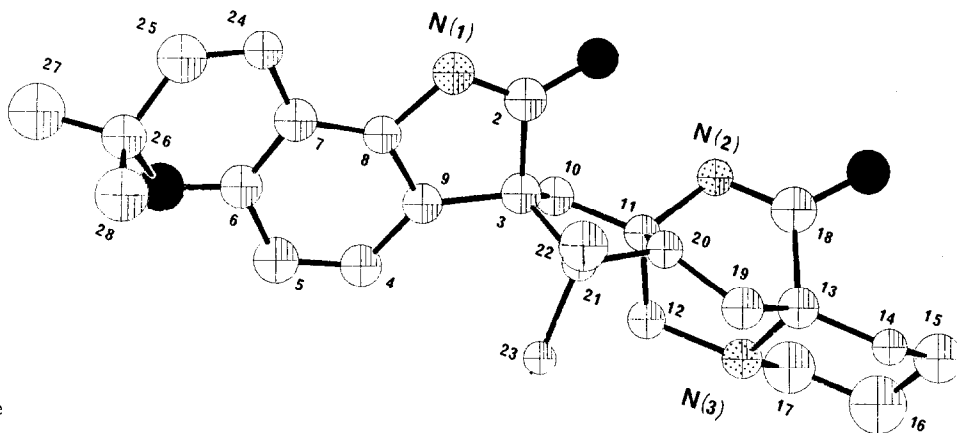
	<u>1</u>	<u>2</u>	<u>3</u> <sup>**</sup>
C-2	183.3 s	183.1	185.5
C-3	60.6 s	61.4	61.0
C-4	120.2 d	120.4	124.4
C-5	114.9 <sup>a</sup> d	115.0 <sup>a</sup>	109.4
C-6	146.2 s	146.1	153.1
C-7	132.8 <sup>b</sup> s	132.6 <sup>b</sup>	105.6
C-8	135.4 <sup>b</sup> s	135.4 <sup>b</sup>	137.8
C-9	124.9 s	124.9	121.2
C-10	37.0 t	39.9	39.6
C-11	63.1 <sup>c</sup> s	61.4 <sup>c</sup>	61.2
C-12	61.5 t	61.4	61.4
C-13	64.2 <sup>c</sup> s	63.1 <sup>c</sup>	62.6
C-14	31.7 t	31.1 <sup>d</sup>	30.8
C-15	20.7 t	20.9	20.7
C-16	25.9 t	25.9	25.7
C-17	54.5 t	54.6	54.5
C-18	173.9 s	177.1	177.2
C-19	31.7 t	31.8 <sup>d</sup>	31.5
C-20	52.9 s	54.2	54.3
C-21	46.5 t	46.7	46.3
C-24	139.2 d	139.1	116.3
C-25	117.2 <sup>a</sup> d	117.8 <sup>a</sup>	131.4
C-26	79.8 s	79.8	76.8
26-Me	29.9 ; 29.9 q	30.1, 29.7	27.8, 27.8
21-Me	20.7 ; 23.8 q	23.9, 20.6	23.7, 20.1
N-Me	26.4 q		

\*\* No Off Res.



Scheme 1

closely related to that of 1 and showed the presence of an indolinone ( $\delta$  183.1) and of an amide function ( $\delta$  177.1). The structure of 2 was further supported by its mass spectrum which showed abundant fragments ions at  $m/z$  420, due to the loss of  $\text{NHCO}$ , and at  $m/z$  177.1480 ( $\text{C}_{12}\text{H}_{19}\text{N}$ , 29 %) and at  $m/z$  149.1195 ( $\text{C}_{10}\text{H}_{15}\text{N}$ , 100 %) which can be attributed to ions (I) and (II), respectively (scheme 1). These ions probably arise by cleavage of the 3,21 carbon bond via a McLafferty rearrangement, loss of the elements of fulminic acid ( $\text{HNCO}$ ) and fragmentation mainly directed by the charge-stabilizing nitrogen atom [N(3)].



Figure

Molecular structure of marcfortine C

Marcfortine C 3 crystallized from ethylacetate as colorless prisms, m.p. 264-267°,  $[\alpha]_D^{22} - 64.4^\circ$  (c 1.1,  $\text{CHCl}_3$ ); u.v.  $\lambda_{\text{max}}$  246 nm ( $\epsilon$  11490). It has the composition  $\text{C}_{27}\text{H}_{33}\text{N}_3\text{O}_3$  ( $M^+ 447$ ) which in conjunction with its  $^1\text{H}$ -n.m.r. spectrum (Table I) indicated that 3 differs from marcfortine B 2 by the presence of singly oxygenated six membered ring located on the tryptophan moiety. The complete structure of marcfortine C 3 was provided by single crystal X-ray analysis. Crystal data: Monoclinic, space group  $P2_1$ ,  $a = 25.274$  (4),  $b = 8.238$  (3),  $c = 13.084$  (3) Å,  $\beta = 91.56$  (6)° and  $Z = 4$ . 4377 (2396  $> 2\sigma$ ) independent reflections were scanned on a four-circle automatic diffractometer using graphite-monochromatized  $\text{Cu-K}\alpha$  radiation. The structure was solved using the Patterson search method <sup>4</sup> with a selected starting set of coordinates from the marcfortine A ring system <sup>1</sup>. The initial rigid model contained 21 atoms. The best figure of merit was the starting point of Fourier recycling procedures which converge to  $R = 26\%$  and revealed the whole set of the 66 atoms of the two independent molecules of the asymmetric unit. Isotropic refinement <sup>5</sup> was conducted to  $R = 13\%$  with hydrogen atoms introduced at their theoretical places. No anisotropic refinement was performed because of the paucity of the data.

The molecular structure of marcfortine C 3 is shown in the Figure. The  $^1\text{H}$  n.m.r. and  $^{13}\text{C}$  n.m.r. (Table I and II) spectra were consistent with

this structure. The electron impact mass spectrum of 3 displayed a fragment ion at  $m/z$  404 (due to the loss of  $\text{NHCO}$ ) and also showed abundant fragment ions in the medium mass range at  $m/z$  177 and 149 (base peak) (see scheme 1).

The similar mass spectral fragmentation patterns of marcfortines B and C and the virtual identical resonances from 0.81 to 3.72 ppm in their  $^1\text{H}$  n.m.r. spectra (Table I) strongly supported the proposed structure 2 for marcfortine B.

Marcfortines A, B and C seem to be biogenetically derived from two units of dimethylallyl-pyrophosphate, pipercolic acid and tryptophan. The loss of the carbonyl oxygen of tryptophan in the supposed formed dioxopiperazine ring has been found previously in fungal alkaloids <sup>6</sup>. The formation of the six membered ring on the tryptophan unit in marcfortine C probably involves, as in the seven membered rings of marcfortine A and B, an epoxy C-6 reverse O-isopentenyl intermediate which after transposition to the corresponding aldehyde undergoes intramolecular nucleophilic attack by the aromatic ring.

Quite recently <sup>7</sup>, the structure of paraherquamide from Penicillium paraherquei has been reported. It also has a seven membered ring located on the tryptophan unit and differs from marcfortine A 1 <sup>1</sup> by the nature of the second aminoacid (2-hydro-2-methyl proline instead of pipercolic acid).

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